

COLLISION-INDUCED ROTATIONAL EXCITATIONS OF INTERSTELLAR
MOLECULES DUE TO He AND H₂

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ABSTRACT. An Effective Straight-line Trajectory (EST) approach by introducing a parameter RX has been proposed¹ for computations of collision-induced rotational line widths (HWHM) and excitation rates in case of atom-molecule systems under the frame work of Smith, Giraud and Cooper² (1976) and molecule-molecule systems under the frame work of normalized semi-classical perturbative approach³. An optimised parameter RX, which is a measure of significance of the curved trajectories of the colliding molecules, can be determined from the temperature dependence of collision-induced line widths. The EST approach has been tested for HCl-Ar system and further applied to X-He and X-H₂ systems of interstellar interest, where X represents interstellar molecules CO, OCS and HCN. The results are given in Table I and II.

The conclusions drawn from the application of EST approach for computations of collision-induced rates are as follows:

a) The collision-induced rotational cross-sections using EST approach are within 5% to 16% with those of Smith et al² and within 0.6% to 14% with those of Robert and Bonamy⁴, in case of HCl-Ar system.

b) The RX dependence of the rates in case of atom-molecule collisions is found to be more sensitive than molecule-molecule collisions.

c) The optimum trajectory corresponds to the optimum value of RX which may be determined using experimental informations. In this study the variation of line width with temperature has been used.

d) Information about the rates with perturber in excited rotational states ($J_2 \neq 0$) can also be determined which is very interesting since rotational population of interstellar molecules is found to be dependent on ortho and para states of interstellar H₂.

e) The computer time required for EST method is only about 20% more than that required for straight-line trajectory approximation for computations of the rates. Hence due to its simple and non expensive nature the EST approach is expected to be more useful in the field.

f) The EST approach is not good for close collisions. The approximations in the potential surface does not allow to take into account the excitations with $\Delta J > 2$. The translational energy has been treated classically hence the approach can not be used at very low temperatures ($T < 50K$).

ACKNOWLEDGEMENTS

We are thankful to the Indian Space Research Organisation (ISRO), Bangalore for financial support.

Table I : Collision-induced rates ($\text{cm}^3 \text{ s}^{-1} \times 10^{10}$) for molecule-molecule systems using EST approach.

| J_i J_f | i | CO-H ₂ | | | OCS-H ₂ | | | HCN-H ₂ | | |
|----------------|-----|-------------------|-------|-------|--------------------|-------|-------|--------------------|--------|-------|
| | | 50°K | 100°K | 250°K | 50°K | 100°K | 250°K | 50°K | 100°K | 250°K |
| 1 | 1-0 | 0.409 | 0.292 | 0.120 | 4.102 | 3.492 | 2.230 | 9.888 | 9.601 | 9.469 |
| | 2-1 | 0.499 | 0.359 | 0.146 | 4.276 | 3.730 | 2.452 | 10.311 | 10.030 | 9.918 |
| 2 | 1-0 | 0.300 | 0.210 | 0.087 | 3.743 | 3.094 | 1.860 | 9.015 | 8.942 | 8.898 |
| | 2-1 | 0.365 | 0.257 | 0.105 | 3.952 | 3.309 | 2.072 | 9.557 | 9.359 | 9.301 |
| 3. | 1-0 | 0.280 | 0.196 | 0.082 | 3.647 | 3.016 | 1.776 | 8.822 | 8.828 | 8.791 |
| | 2-1 | 0.343 | 0.240 | 0.099 | 3.876 | 3.235 | 1.999 | 9.326 | 9.217 | 9.167 |
| 4 | 1-0 | 0.273 | 0.191 | 0.080 | 3.609 | 2.978 | 1.745 | 8.751 | 8.787 | 8.751 |
| | 2-1 | 0.335 | 0.234 | 0.096 | 3.846 | 3.201 | 1.971 | 9.251 | 9.165 | 9.118 |
| 5 | 1-0 | 0.269 | 0.188 | 0.079 | 3.590 | 2.958 | 1.730 | 8.731 | 8.767 | 8.733 |
| | 2-1 | 0.331 | 0.231 | 0.095 | 3.832 | 3.184 | 1.957 | 9.215 | 9.141 | 9.095 |

Table II : Line width parameters $\Delta\nu$ (MHz/torr) for atom-molecule systems using EST approach.

| J_i J_f | i | CO-He | | | OCS-He | | | HCN-He | | |
|----------------|-----|-------|-------|-------|--------|-------|-------|--------|-------|-------|
| | | 50°K | 100°K | 300°K | 50°K | 100°K | 300°K | 50°K | 100°K | 300°K |
| 1-2 | | 7.007 | 4.297 | 1.576 | 12.910 | 5.730 | 1.500 | 11.541 | 7.099 | 3.567 |
| 2-3 | | 6.963 | 4.209 | 1.571 | 12.335 | 5.619 | 1.522 | 8.718 | 5.523 | 2.350 |

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