

## Nanoscale Quantification of Jahn-Teller Distortion in LaMnO<sub>3</sub>

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Cooperative Jahn-Teller distortions (CJTD), which appear in the octahedral sites of some transition metal ions (i.e. Cu<sup>2+</sup> and Mn<sup>3+</sup>), is a phenomenon related to many interesting physical properties, including charge ordering, spin ordering and orbital ordering [1, 2]. For example, in LaMnO<sub>3</sub>, the electron-phonon interaction causes the Jahn-Teller splitting of e<sub>g</sub> levels, where a Q<sub>2</sub> octahedral distortion is typically observed: two Mn-O bonds shorten, and two Mn-O bonds lengthen [3].

Experimentally, a precise determination of the octahedral bonding parameters in oxides can be achieved by Rietveld refinement of neutron and X-ray diffraction techniques [4-6]. However, these techniques probe volume averaged properties of materials. In order to study the local properties of materials, such as interface coupling of octahedral connectivity in perovskite oxide heterostructures and the topological hall effects driven by octahedral tilting in ultrathin films aberration corrected scanning transmission electron microscopy (STEM) offers the highest spatial resolution [7, 8].

Due to the relatively weak scattering cross-section of oxygen (O), techniques such as annular bright field (ABF) imaging, electron energy loss spectra (EELS) and differential phase contrast (DPC) imaging are necessary to image the O columns in real space [9-11]. However, rotation of the octahedra in complex oxides can make it challenging to image O columns precisely and potential image distortion caused by probe instabilities and sample drift can further complicate the situation. For example, as shown in Fig. 1a, certain oxygen sites are co-aligned with some of the heavy atom positions in the [110] projection. While post-processing procedures can allow the octahedral rotation types to be deduced in certain cases, the magnitude of the bond-distance changes in the oxygen octahedra is a challenge in even the most advanced STEM instruments. For example, Mn-O bond lengths in LaMnO<sub>3</sub> differ by 6-27 pm due to CJTD; whereas, without considering CJTD, they only differ by approximately 7 pm.

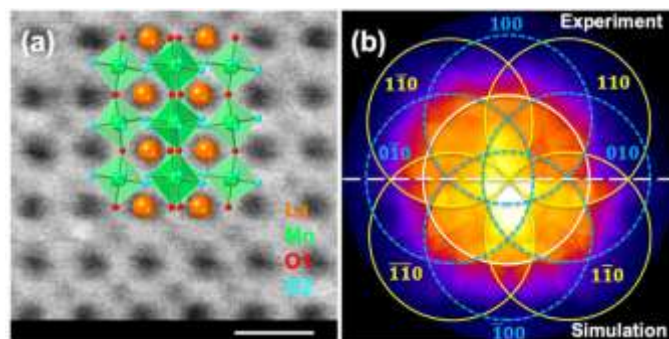
An alternative technique to investigate the structural origins of the properties of nanoscale perovskites with distorted octahedra is position-averaged convergent beam electron diffraction (PACBED), which has been proven to be sensitive to structural information, such as thickness, distortion and polarity [12]. As shown in Fig. 1b, the simulated PACBED is in good agreement with experimental results for a sample thickness of approximately 27 nm. By comparing experimental and simulated results, octahedral rotation in perovskites can be quantitatively measured [13].

In this work, we propose an approach to precisely detect the magnitude of CJTD using PACBED. The picometer-size CJTD-introduced change in bond distance in LaMnO<sub>3</sub> can be detected experimentally and verified by simulation. In Fig. 2, the minimum normalized X<sup>2</sup> shows the best agreement between simulated PACBED and experimentally derived values [4] when considering a combination of octahedral rotations at different bonding ratios (Mn-O<sub>2</sub>(short)/Mn-O<sub>2</sub>(long)). The optimized fit appears when the bonding ratio is the closest to the experimental values, as indicated by the arrow in Fig. 2a. These data show that octahedral distortions due to CJTD are experimentally detectable when verified by simulation, helping to

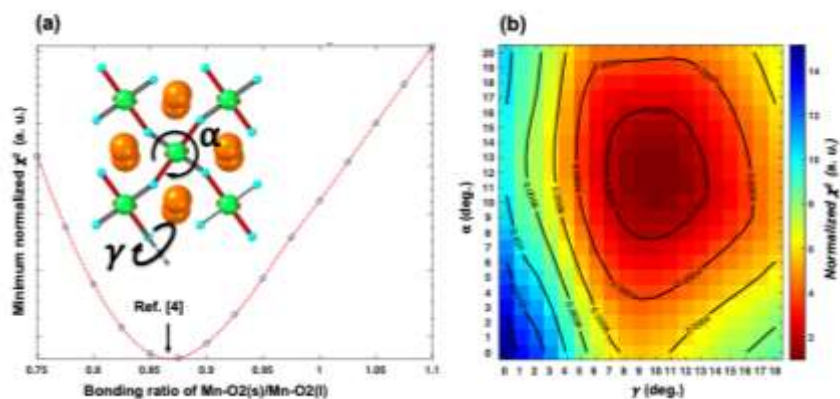
investigate local CJTD-related distortions and their effect on physical properties.

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**Figure 1.** (a) Bright field STEM image of  $\text{LaMnO}_3$  along the  $[110]$  after non-rigid registration with  $\text{Pbnm}$  crystal model overlay. Scale bar is 1 nm. (b) Comparison of experimentally measured and simulated PACBED of the  $\text{LaMnO}_3$   $[001]$  with 27 nm thickness.



**Figure 2.** Quantification of the effects of (a) Jahn-Teller distortions and (b) octahedral tilts in simulated PACBED of  $\text{LaMnO}_3$ .