# Developing biomarkers to estimate manure ammonia emissions from dairy cattle

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## Animal Research Paper

## Abstract

This research aimed to develop biomarkers for estimating ammonia ($NH_3$) emissions from dairy cattle manure over a 15-day *in vitro* incubation system. To generate different levels of $NH_3$ emissions, the experiment utilized four manure experimental groups: 1 urinary nitrogen (U) to 1 faecal nitrogen (F) ratio (CT), 2 U to 1 F ratio (2U1F), and CT and 2U1F with lignite application (CT + L and 2U1F + L, respectively). The addition of lignite to ruminant manure aimed to enhance environmental sustainability through its beneficial properties. Three biomarkers, nitrogen (N) isotopic fractionation ($\delta^{15}N$), N: potassium (K) ratio, and N: phosphorus (P) ratio, were investigated. Manure $\delta^{15}N$ increased linearly when $NH_3$ emission increased in CT and 2U1F groups ($R^2 = 0.79$ and 0.90, respectively; $P \leq 0.001$), while manure N: P decreased when $NH_3$ emission increased in CT + L and 2U1F + L groups ($R^2 = 0.73$ and 0.85, respectively; $P \leq 0.001$). No useful relationship was found between N: K and $NH_3$ emission, apart from in 2U1F group ($R^2 = 0.84$; $P \leq 0.001$). The experiment found manure $\delta^{15}N$ and N: P are complementary biomarkers to predict $NH_3$ emissions, from non-lignite and lignite groups, respectively.

## Introduction

Livestock system relies heavily on nitrogen (N) as a crucial nutrient for production (i.e., meat, milk, etc.). However, a significant portion, approximately 70%, of the N from feed consumed by dairy cows is excreted as urine and faeces, rather than being converted into saleable products (Yan *et al.*, 2006; Kebreab *et al.*, 2010; Cheng *et al.*, 2013). This excretion, when urine encounters faeces, leads to the production of ammonia ($NH_3$) gas through the reaction of urinary urea with the urease enzyme present in faeces. Livestock, particularly cows, contribute up to 40% of $NH_3$ emissions of agricultural origin, primarily originating from manure (Chen *et al.*, 2015; Brandani *et al.*, 2023). Such emissions pose environmental threats, causing air and water pollution (Bouwman *et al.*, 2002; Todd *et al.*, 2006), and negatively impacting ecosystems and public health (WHO, 2021). The volatilized $NH_3$ can further contribute to the formation of nitrous oxide ($N_2O$), a potent greenhouse gas (Chen *et al.*, 2015), resulting in nutrient loss and reduced manure value as a fertilizer source, affecting dairy cow productivity and farm profitability (Stott and Gourley, 2016). Consequently, accurate quantification and research into strategies for mitigating $NH_3$ emissions are crucial for sustainable livestock production.

Quantifying $NH_3$ emissions is challenging due to its rapid reactivity in the atmosphere. Direct quantification methods involve chemical reactions and include acid traps, detector tubes, electrochemical sensors, passive gas absorption, denuders, optical absorption techniques and chemiluminescence techniques (Hristov *et al.*, 2011). However, the application of these methods faces challenges at larger scales due to logistical complexities, geographic requirements and high labour and maintenance costs. Therefore, indirect methods, such as biomarkers (manure N: phosphorus [P] ratio, N: potassium [K] ratio, and N isotopic fractionation [$\delta^{15}N$]), offer alternatives (Hristov *et al.*, 2009). However, these emerging indirect, biomarker-based methods were not explored widely under diverse manure production conditions.

Thus, this experiment aimed to add value to the literature by testing the hypothesis that changes in manure $\delta^{15}N$, N: P, and N: K can serve as indicators for estimating $NH_3$ emissions, with a wide range of $NH_3$ emissions generated from manipulation of manure N composition and application of lignite as a known $NH_3$ mitigation product.

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## Material and Methods

The experiment was approved by the Animal Ethics Committee from the Faculty of Science, The University of Melbourne, Australia (No. 1914778.1).

### Farm location, milking and feeding system

The experiment was conducted at Dookie dairy farm (−36.4°S, 145.7°E. and 140 m above sea level). The farm has a total area of 126 ha and operates a grazing/feed pad feeding system. The farm has an Automatic Milking System which has a capacity of milking 180 mixed-age dairy cattle (AMS; Lely Astronaut; Lely, Maassluis, The Netherlands), coupled with an automatic weighing and pellet feeding system. A total of 20 lactating cattle (average live weight = 608.1 ± 69.64 kg; average lactation day = 207.8 ± 118.53 d; mean ± SD) were selected from the mixed age Holstein Friesians cattle herd at 9:00 am on 15th August 2019.

### Feed sampling and nutritive value analysis

The feed samples were collected at 9:00 am daily from the farm between 12th and 14th of August 2019, three days prior to the start of urine and faeces samplings. The quantity of pellets offered, refused, and consumed by each dairy cattle per time when they visited the robots was recorded by AMS. The pellets were directly sampled through AMS. Barley hay was sampled by taking one core sample from each bale used during the experimental period. A total of 15 fresh oat crop samples (cut full height above ground) were randomly sampled from the paddock on 12th August 2019. The same method was used to sample ryegrass-based pasture on the 13th and 14th of August 2019. The collected feed samples were stored at −20°C in the Dookie Campus Science Laboratory. All frozen feed samples were then bulked per sample type and directly sent with an icebag to the New South Wales Department of Primary Industries (NSW DPI), Australia, for analysing DM, crude protein, DM digestibility (DMD), digestible organic matter in DM and estimated metabolizable energy (ME) using a Near-Infrared Reflectance Spectroscopy (NIR systems model 6500, Foss, Hillerød, Denmark).

### Metabolizable energy measurement and diet intake estimation

The ME required for cattle maintenance ($ME_{maintenance}$), activity ($ME_{activity}$), lactation ($ME_{lactation}$), diet ME intake ($ME_{diet\ intake}$), diet DM intake (DMI) and ME concentration of the diet ($ME_{diet}$) were calculated following the published equations (Eqns (1)–(6)) following Nicol and Brookes (2007):

$$ME_{maintenance}(MJ/d) = 0.56\ MJ\ ME/kg \\ \times Live\ weight^{0.75}(LW;\ kg) \quad (1)$$

$$ME_{activity}(MJ/d) = LW(kg) \times (0.0026 \times Hkm(km/d) \\ + 0.028 \times Vkm(km/d)) \div K_m \quad (2)$$

where Hkm was horizontal walking distance (km/d) and Vkm was vertical climbed distance (km/d) that were estimated from google map distance with identified daily grazing zone location; $K_m = ME_{diet}$ (MJ/kg DM) × 0.02 + 0.5.

$$ME_{lactation}(MJ/d) = 1.1 \times milk\ yield(MY)\ (kg/d) \\ \times Net\ energy(NE;\ MJ/kg) \div k \quad (3)$$

where NE (MJ NE/kg) = 0.376 × Fat concentration of milk + 0.209 × Protein concentration of milk + 0.976; $k$ is the efficiency with which ME is utilized, $k = ME_{diet}$ (MJ/kg DM) × 0.02 + 0.4.

The fat and protein analyses were performed by infrared spectroscopy method.

$$ME_{diet\ intake}(MJ/d) = ME_{maintenance}(MJ/d) \\ + ME_{activity}(MJ/d) \quad (4) \\ + ME_{lactation}(MJ/d)$$

$$DMI(kg\ DM/d) = Offered\ barley\ hay(kg/d) \times (1 - Wastage(\%)) \\ \times DM\ percentage\ in\ barley\ hay(\%) \\ + Commercial\ cereal\ grain\text{-}based \\ pellets\ intake(kg/d) \\ \times DM\ percentage\ in\ pellets(\%) \\ + Grazing\ intake(kg/d) \\ \times DM\ percentage\ in\ oat\ crops\ or\ pasture(\%) \quad (5)$$

where grazing intake (kg/d) = Oat crops intake (kg/d) for day 1; Pasture intake (kg/d) for days 2 and 3.

$$ME_{diet}(MJ/kg\ DM) = ME_{diet\ intake}(MJ/d) \div DMI(kg\ DM/d) \quad (6)$$

The ME for grazing intake (oat crops for day 1 and pasture for days 2 and 3), pellets intake, and barley hay intake were calculated by following equations (Eqns (7)–(9)):

$$ME_{barley\ hay\ intake}(MJ/d) = Offered\ barley\ hay(kg/d) \\ \times (1 - Wastage(\%)) \\ \times DM\ percentage\ in\ barley\ hay(\%) \\ \times ME_{barley\ hay}(MJ/kg\ DM) \quad (7)$$

$$ME_{pellets\ intake}(MJ/d) = Commercial\ cereal\ grain\text{-}based\ pellets \\ intake(kg/d) \\ \times DM\ percentage\ in\ pellets(\%) \\ \times ME_{pellets}(MJ/kg\ DM) \quad (8)$$

$$ME_{grazing\ intake}(MJ/d) = Oat\ crops\ intake\ for\ 1\ day(kg/d) \\ \times DM\ percentage\ in\ oat\ crops(\%) \\ \times ME_{oat\ crops}(MJ/kg\ DM);\ Pasture \\ intake\ for\ days\ 2\ and\ 3(kg/d) \\ \times DM\ percentage\ in\ pasture(\%) \\ \times ME_{pasture}(MJ/kg\ DM) \quad (9)$$

Besides,

$$ME_{diet\ intake}(MJ/d) = ME_{barley\ hay\ intake}(MJ/d) \\ + ME_{pellets\ intake}(MJ/d) \quad (10) \\ + ME_{grazing\ intake}(MJ/d)$$

The grazing intake could be calculated using (Eqns (1) to (10)), then $ME_{activity}$, $ME_{lactation}$, $ME_{diet\ intake}$, $ME_{grazing\ intake}$, DMI and $ME_{diet}$ could be calculated based on grazing intake.

## Urine and faeces sampling and analysis

A 300 ml sample of urine was collected from each dairy cattle, following vulva stimulation for a maximum of 5 min. If urination did not occur within 5 min, the dairy cattle were released, and sample collection was attempted with the next dairy cattle in the queue. A total of 6 l urine from 20 cattle was collected and well-mixed in a bucket. To reduce $NH_3$ emissions from urine samples, a container with ice was placed below the urine collection bucket. After each urine sampling, 300 g of faeces sample was collected from rectal grab. A total of 6 kg of faeces samples from 20 cattle was collected and mixed in a bucket.

For UN% and FN% analysis, two 5 ml of well-mixed urine samples and one 5 g of faeces sample were subsampled immediately into 15 ml test tubes, and one drop of concentrated $H_2SO_4$ (98%) was added into each tube for sample acidification (with a targeted sample pH < 4). All samples were stored in a freezer at −20°C at Dookie Campus Science Laboratory. One urine and one faeces subsample were pre-processed and analysed via Dumas Combustion on a LECO Trumac CN (TruMac CNS Macro Analyzer, LECO Corporation, Australia) at a furnace temperature of 1250°C. Results of UN% and FN% were reported as w/w concentration. Another urine subsample was analysed for creatinine concentration (Cobas Integra 400 plus, Roche Diagnostics, Australia).

## Nitrogen balance calculation

The N balance (NB) was calculated by the following equations (Eqn. (11)) according to Spanghero and Kowalski (1997):

$$NB(g/d) = NI(g/d) - UN(g/d) - FN(g/d) - \text{Milk } N(g/d) \tag{11}$$

where NI was N intake, which was calculated from dietary N concentration and DMI; the UN and FN were urinary N and faecal N; the milk N, which was calculated with individual dairy cattle MY and milk protein concentration (provided by Lely robotic milking system) ÷ 6.38, 6.38 was the conversion factor of N values from milk protein values.

The NI, UN, FN and MN were calculated as following equations (Eqns (12) to (20)):

$$NI_{\text{diet intake}}(g/d) = NI_{\text{pellets}}(g/d) + NI_{\text{barley hay}}(g/d) + NI_{\text{grazing}}(g/d) \tag{12}$$

$$
\begin{aligned}
NI_{\text{pellets}}(g/d) &= \text{Crude protein concentration in pellets} \\
&\quad (\% \text{ on a DM basis}) \\
&\quad \div 6.25 \times \text{Commercial cereal grain-based pellets} \\
&\quad \text{intake}(kg/d) \times \text{DM percentage in pellets}(\%) \\
&\quad \times 1000, \text{ where factor } 6.25 \text{ was for the conversion} \\
&\quad \text{of } N \text{ values from protein values.}
\end{aligned} \tag{13}
$$

$$
\begin{aligned}
NI_{\text{barley hay}}(g/d) &= \text{Crude protein concentration in barley hay} \\
&\quad (\% \text{ on a DM basis}) \\
&\quad \div 6.25 \times \text{Offered barley hay daily}(kg/d) \\
&\quad \times (1 - \text{Wastage}(\%)) \\
&\quad \times \text{DM percentage in barley hay}(\%) \times 1000
\end{aligned} \tag{14}
$$

$$
\begin{aligned}
NI_{\text{grazing}}(g/d) &= \text{Crude protein concentration in oat crops} \\
&\quad (\% \text{ on a DM basis}) \\
&\quad \div 6.25 \\
&\quad \times \text{Oat crops intake for day 1}(kg/d) \\
&\quad \times \text{DM percentage in oat crops}(\%) \\
&\quad \times 1000; \text{ Crude protein concentration in pasture} \\
&\quad (\% \text{ on DM basis}) \div 6.25 \\
&\quad \times \text{Pasture intake}(kg/d) \text{for day 2 and 3} \\
&\quad \times \text{DM percentage in pasture}(\%) \times 1000
\end{aligned} \tag{15}
$$

where,

$$UN(g/d) = UN\% \times UW(kg/d) \times 1000 \tag{16}$$

where UW is the weight of urine excreted daily from dairy cattle, and

$$
\begin{aligned}
UW(kg) &= 21.9(mg/kg) \times LW(kg) \\
&\quad \div \text{Urine creatinine}(mg/kg)
\end{aligned} \tag{17}
$$

$$
\begin{aligned}
FN(g/d) &= FN\% \times DMI(kg/d) \\
&\quad \times (1 - \text{Dietary DM digestibility}(DMD \%)) \\
&\quad \times 1000
\end{aligned} \tag{18}
$$

$$
\begin{aligned}
&\text{Dietary DM digestibility}(DMD \%) \\
&= \left( \begin{array}{l} \text{Commercial cereal grain-based pellets intake}(kg/d) \\ \times \text{DM percentage in pellets}(\%) \times \text{DMD of pellets}(\%) \\ + \text{Offered barley hay}(kg/d) \times \left( \begin{array}{l} 1 - \text{Wastage} \\ (\%) \end{array} \right) \\ \times \text{DM percentage in barley hay}(\%) \times \text{DMD of barley} \\ \text{hay}(\%) + \text{Grazing intake}(kg/d) \times \text{DM} \\ \text{percentage in oat crops or pasture}(\%) \times \text{DMD of} \\ \text{oat crops or pasture}(\%) \end{array} \right) \\
&\quad \div DMI(kg/d)
\end{aligned} \tag{19}
$$

$$
\begin{aligned}
\text{Milk } N(g/d) &= \text{Milk protein concentration } \% \div 6.38 \\
&\quad \times MY(kg/d) \times 1000, \\
&\quad \text{Where factor } 6.38 \text{ was for the conversion of } N \\
&\quad \text{values from milk protein values}
\end{aligned} \tag{20}
$$

## Ammonia emissions measurement using dairy cattle manure through in vitro incubation

### Lignite properties and application preparation

Raw lignite (Latrobe Valley, Victoria, Australia) was ground by a pestle and mortar and then passed through a sieve (approximately 500 μm in diameter). A total of 2.1 kg of lignite powder was prepared, and one 10 g of subsample was dried in the oven (65°C) for three days to determine the DM of lignite. One 10 g of lignite subsample was sent to the University Melbourne TrACEES Platform

for the analysis of total N concentration and N isotope fractionation ($\delta^{15}$N) using a Thermo Flash 2000 HT (elemental analyser, Thermo Scientific, Australia) paired with a Thermo Delta V Advantage (mass spectrometer, Thermo Scientific, Australia) (CF – IRMS). The results were calculated using thermal conductivity detector (TCD) output, which was calibrated against acetanilide (Thermo Fischer, Bremen, Germany), and reported as weight/weight concentration. The analysis of lignite pH and electrical conductivity (EC) was performed based on sections 3A1 and 4A1 in the soil chemical methods book published by Rayment and Lyons (2011).

### Incubation system and experimental design
Eight incubation systems were constructed at the Dookie Campus Science Laboratory, at the University of Melbourne. The incubation system was adopted from Lee *et al.* (2011) study. Each system consisted of 5 components: (1) an air pump (Stella 110, Aqua One, Australia), which propelled the air into the system; (2) an airflow meter (LZB-3WB 0.15–1.5 l/min, Darhor, China), which controlled and monitored the airflow rate; (3) a 900 ml clip water container (12.5 cm height, 13.5 cm width, and 10.5 cm length, Kmart, Australia), which was filled with water (500 ml) to provide moisture and to keep the surface of the manure from drying out during the incubation period; (4) a 2.3 l clip manure incubation container, which was made of polypropylene and silicone (16.6 cm height and 16.4 cm diameter, Kmart, Australia); and (5) an acid trap composed of a 500-ml volumetric flask (Flask Erlenmeyer 500 ml 29/30, Quickfit, England), which contained 0.5 M sulphuric acid to capture the emitted $NH_3$ from the manure incubation container (Supplementary file). A clear vinyl tube (10 mm diameter, Pope, Australia), made of plasticized polyvinyl chloride compound, was used to connect each component of the incubation system. A universal tap adaptor (12 mm diameter, Pope, Australia) was used to secure and provide a leak-free connection between containers and tubes. A leakage test was performed for each incubation system before experimenting. A detergent (Dishwashing liquid, Earth choice, Ferntree Gully, Victoria, Australia) and water solution was applied to each connection point after the air pump was on. This was done to detect leaks, which would cause bubbles to form if present.

The incubation experiment was conducted at room temperature at the Parkville Campus, the University of Melbourne, Melbourne. The experimental design included one four treatment groups: 1 urinary nitrogen (U) to 1 faecal nitrogen (F) ratio (CT), 2 U to 1 F ratio (2U1F), and CT and 2U1F with 250 g lignite application (CT + L and 2U1F + L, respectively). The applied raw lignite was sourced from Latrobe Valley, Victoria, Australia. The incubation experiment had two trial periods, and each included two replications per treatment, with a total of four replications achieved per group for data analysis. The first period was conducted from 18th October to 2nd November 2019, and the second period was conducted from 15th November to 30th November 2019. Each period lasted for 15 days. Each incubation system contained 600 g reconstructed manure (Table 1) based on the pre-determined UN% and FN% (LECO Trumac CN method, TruMac CNS Macro Analyzer, LECO Corporation, Australia) to achieve a manure UN to FN ratio of 1 to 1 for CT and CT + L, and 2 to 1 for 2U1F and 2U1F + L. Further, 250 g lignite powder was added into CT + L and 2U1F + L and mixed with manure. To reconstruct the manure, frozen urine and faeces were thawed for 24 h at room temperature and weighed according to Table 1. They were then mixed using a glass rod for 30 s before being added to

**Table 1.** The input of urine, faeces, and lignite in the experiment

|  | Treatment groups | | | |
|---|---|---|---|---|
|  | Control (CT[a]) | 2U1F[b] | CT + L[c] | 2U1F + L[d] |
| Urine input (g) | 192 | 291 | 192 | 291 |
| Faeces input (g) | 408 | 309 | 408 | 309 |
| UN (g) | 1.67 | 2.53 | 1.67 | 2.53 |
| FN (g) | 1.67 | 1.27 | 1.67 | 1.27 |
| Lignite (g) | 0 | 0 | 250 | 250 |

[a]CT used a manure with UN:FN = 1:1.
[b]2U1F used a manure with UN:FN = 2:1.
[c]CT + L used a manure with UN:FN = 1:1 + 250 g lignite application.
[d]2UIF + L used a manure with UN:FN = 2:1 + 250 g lignite application; UN:FN = urinary nitrogen: faecal nitrogen.

the container for incubation. The acid trap flask contained 500 ml of 0.5 M $H_2SO_4$ daily. The chemical reaction in the aid trap is expressed by following the equation (Eqn. 21):

$$2NH_3 + H_2SO_4 = (NH_4)_2SO_4, \text{ where } (NH_4)_2SO_4 \text{ is ammonium sulphate} \tag{21}$$
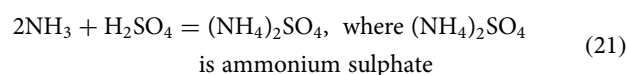
### Sampling and analysis of fresh manure and acid trap samples

#### Manure samples from *in vitro incubator*
During each incubation period, manure samples from each treatment were collected at 10:00 once a day. For each collection, the air pump was turned off, and an incubation container was opened for less than one minute to collect manure using a calibrated plastic straw from five randomly selected spots. Five manure samples were then bulked to provide ∼ 15 g of manure sample per replication daily for analysis. After the manure sample was collected, the container was closed, and the air pump was turned on to continue incubation. Each collected manure sample was freeze-dried by a freeze dryer (GAMMA 1–16 LSC plus, Martin Christ Freeze Dryer, Germany) for five days, then weighed to determine the DM. Each freeze-dried manure sample was then ball-milled using a tissue lyser (TissueLyser II, Qiagen, Germany) prior to chemical analysis. The $\delta^{15}$N and total N of manure samples were analysed by the CF – IRMS. Both data were calculated using TCD output, calibrated against acetanilide (Thermo Fischer, Bremen, Germany), and reported as weight/weight concentration. Manure P and K samples were analysed using a Perkin Elmer 8300 DV ICP-OES (Syngystix v3.0, Perkin Elmer, China) relative to a 5% nitric acid sample matrix. Data were reported as weight/weight concentration.

#### Acid trap sample collection and analysis
A total of 15 ml of acid sample was collected daily from each incubation system and stored frozen (−20°C). The collected acid was used to determine $NH_3$ concentration per 24-h incubation period. The distillation and titration methods were used to determine $(NH_4)_2SO_4$ concentration in acid samples (Vinklárková *et al.*, 2015). The concentration of $(NH_3)$ was calculated following the equation (Eqn. 22):

$$NH_3(g/l) = (NH_4)_2SO_4(g/l) \times 17.03 \times 2/132.14 \tag{22}$$

where 17.03 and 132.14 are the molar mass of $NH_3$ and $(NH_4)_2SO_4$ (g/mol), respectively, factor 2 is the match number from the chemical reaction.

### Statistical analysis

The manure total N, P and K concentration data were corrected for the manure total N, P and K concentrations that would be from samples removed from incubation systems using daily manure total N, P and K concentration data and manure amount. The manure $\delta^{15}$N and $NH_3$ data were corrected for the $\delta^{15}$N and $NH_3$ that would be from samples removed from incubation systems using daily manure $\delta^{15}$N and $NH_3$ data and corrected manure total N.

The Genstat statistical package (16th version, VSN International group) was used to perform statistical analysis in this experiment, including general ANOVA, and linear/power/polynomial regressions. The significance of linear was determined using general ANOVA with manure composition + lignite application + day as treatment structure, and replication/treatment as block structure.

### Results

#### Daily ammonia-nitrogen emission, cumulative ammonia-nitrogen, and manure nitrogen isotope fractionation

In the CT and 2U1F treatments (non-lignite groups), the $NH_3$-N emission decreased over time. The curves of CT + L and 2U1F + L treatments (lignite groups) were the opposite with non-lignite groups, which increased with days. There was a weak relationship between temperature and daily $NH_3$-N emission in four treatments ($R^2 = 0.30$, $P \leq 0.05$; Supplementary file).

The cumulative $NH_3$-N emission increased over time in four treatments ($R^2 = 0.99$, $P \leq 0.001$; Supplementary file). The lignite application reduced the cumulative $NH_3$-N emission by 87.5% compared to CT and CT + L treatments and 82.9% compared to 2U1F and 2U1F + L treatments. The $NH_3$-N divided by manure total N was used to calculate the percentage of manure total N lost as $NH_3$-N. It was found that 80% and 63.4% of the total N loss from manure was lost as $NH_3$-N in the CT and 2U1F treatments, respectively.

The manure $\delta^{15}$N (‰) from the CT and 2U1F treatments from 15 days of dairy manure incubation increased with incubation days ($R^2 = 0.85$ and 0.91, respectively; $P \leq 0.001$). However, there was no relationship between manure $\delta^{15}$N from the CT + L, 2U1F + L treatments and incubation day ($R^2 = 0.07$ and 0.08, respectively; $P > 0.05$).

#### Manure nitrogen to phosphorus ratio and manure nitrogen to potassium ratio

The manure N: P ratio (g/g) from four treatments decreased with incubation days ($R^2 = 0.81$, 0.73, 0.81 and 0.84 in CT, 2U1F, CT + L, and 2U1F + L treatment, respectively; $P \leq 0.001$; Supplementary file). The manure N: P ratio decreased with incubation days in four treatments. In the non-lignite group, the manure N: P ratio from the 2U1F treatment was higher than it was from the CT treatment.

There was a weak relationship between manure N: K ratio and incubation days found in CT treatment ($R^2 = 0.28$, $P \leq 0.001$; Supplementary file), while the relationships were strong in the other three treatments ($R^2 = 0.70$, 0.79, and 0.90 in 2U1F, CT + L, and 2U1F + L treatment, respectively; $P \leq 0.001$). The manure N: K ratio decreased with incubation days in four treatments. Opposite to the manure N: P ratio, the manure N: K ratio from the CT treatment was higher than it was from the 2U1F

treatment. Similar to the non-lignite group, the manure N: K ratio from the 2U1F + L treatment was lower than the CT + L treatment.

#### Cumulative ammonia-nitrogen emission and manure nitrogen isotope fractionation

Figure 1(a) and (b) show the relationships between cumulative $NH_3$-N emission and manure $\delta^{15}$N. In the non-lignite group, there were strong relationships between cumulative $NH_3$-N emission and manure $\delta^{15}$N ($R^2 = 0.79$ and 0.90 in CT and 2U1F treatment, respectively; $P \leq 0.001$). Besides, the relationship between cumulative $NH_3$-N emission and manure $\delta^{15}$N in the 2U1F treatment was stronger than it was in the CT treatment. However, there was no relationship found in the lignite group ($R^2 = 0.02$ and 0.0003 in CT + L and 2U1F + L treatment, respectively; $P \leq 0.001$). The strong relationship between cumulative $NH_3$-N emission and manure $\delta^{15}$N from the non-lignite group (combined CT and 2U1F treatments; $R^2 = 0.83$, $P \leq 0.001$; Fig. 1(c)). However, no relationship was found when combined CT + L and 2U1F + L treatments (Lignite group; $R^2 = 0.02$, $P \leq 0.001$; Fig. 1(d)).

#### Cumulative ammonia-nitrogen emission, and nitrogen to phosphorus and potassium ratios

The cumulative $NH_3$-N emission and manure N: P ratio from CT treatment had a relatively strong relationship ($R^2 = 0.67$, $P \leq 0.001$) while it was moderate from the 2U1F treatment ($R^2 = 0.54$, $P \leq 0.001$; Fig. 2). In the lignite group, the cumulative $NH_3$-N emission and manure N: P had strong relationships ($R^2 = 0.73$ and 0.85 in CT + L and 2U1F + L treatments, $P \leq 0.001$).

The cumulative $NH_3$-N emission and manure N: K ratio from the 2U1F treatment had a strong relationship ($R^2 = 0.84$, $P \leq 0.001$) while it was moderate from the CT treatment ($R^2 = 0.48$, $P \leq 0.001$; Fig. 3). In the lignite group, cumulative $NH_3$-N emission and manure N: K ratio relationship was weak in CT + L treatment ($R^2 = 0.64$, $P \leq 0.001$) and was relatively strong in 2U1F + L treatment ($R^2 = 0.34$, $P \leq 0.001$).

### Discussion

The experimental design drew inspiration from the NB study and previous research by Spanghero and Kowalski (1997), establishing UN and FN outputs as key metrics. In the non-lignite group, $NH_3$-N emissions exhibited a decreasing trend over incubation days, driven by diminishing urinary urea concentration and hydrolysis.

The major $NH_3$-N emission occurred in the last 10 days in lignite groups. This could be caused by lignite reaching the maximum capacity to suppress $NH_3$-N emission. The increasing trend agreed with the results of Impraim et al. (2020). Although Impraim et al. (2020) found the same trend of the daily $NH_3$-N emission and temperature, the weak relationships between temperature and daily $NH_3$-N emission in four treatments ($R^2 = 0.30$) were observed in our experiment. This modest correlation may be attributed to the controlled environmental conditions within the laboratory setting, wherein temperature fluctuations were limited to a narrow range. It is plausible that a broader spectrum of temperature variations would yield a more pronounced relationship. Therefore, expanding the range of temperature fluctuations is anticipated to enhance the strength of the observed association. The manure $\delta^{15}$N from the CT and 2U1F treatments
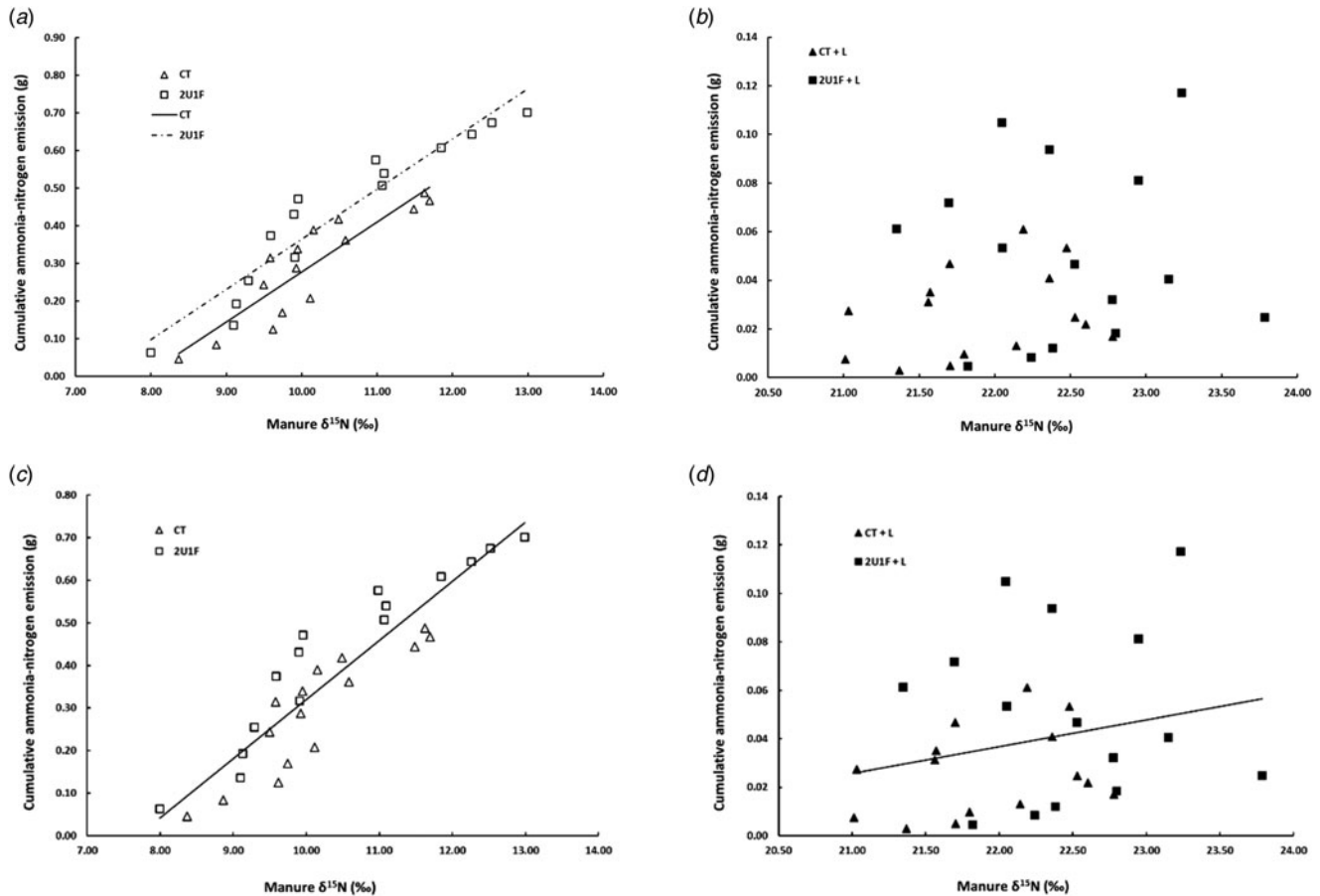
**Figure 1.** Cumulative ammonia-nitrogen emission (g) and manure $\delta^{15}N$ (‰) from 15 days of dairy manure incubation. CT: UN:FN = 1:1; 2U1F: UN:FN = 2:1; CT + L: CT + Lignite; 2U1F + L: 2U1F + Lignite. CT used a manure with UN:FN = 1:1; 2U1F used a manure with UN:FN = 2:1; CT + L used a manure with UN:FN = 1:1 + 250 g lignite application; 2UIF + L used a manure with UN:FN = 2:1 + 250 g lignite application; UN:FN, urinary nitrogen: faecal nitrogen.

had a strong relationship with incubation days, similar relationships were reported in studies by Hristov *et al.* (2009) and Lee *et al.* (2011).

The manure N: P ratio from four treatments had a strong relationship with incubation days ($R^2 = 0.84$, 0.73, 0.81, and 0.84 for CT, 2U1F, CT + L, and 2U1F + L treatments, respectively). This strong relationship was also proved by Moreira and Satter (2006), who stated that the N: P ratio decreased from 6.57 g/g in fresh manure to 2.33 g/g in manure incubated after 30 days. The weak relationship between manure N: K ratio and incubation
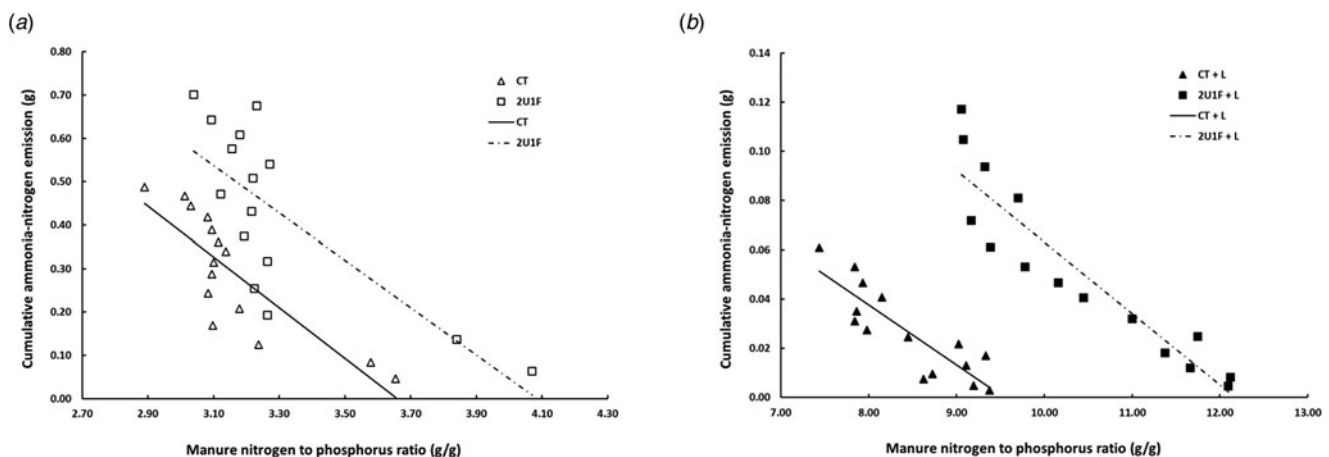


**Figure 2.** Cumulative ammonia-nitrogen emission (g) and nitrogen to phosphorus ratio (g/g) from 15 days of dairy manure incubation. CT: UN:FN = 1:1; 2U1F: UN:FN = 2:1; CT + L: CT + Lignite; 2U1F + L: 2U1F + Lignite. CT used a manure with UN:FN = 1:1; 2U1F used a manure with UN:FN = 2:1; CT + L used a manure with UN:FN = 1:1 + 250 g lignite application; 2UIF + L used a manure with UN:FN = 2:1 + 250 g lignite application; UN:FN, urinary nitrogen: faecal nitrogen.
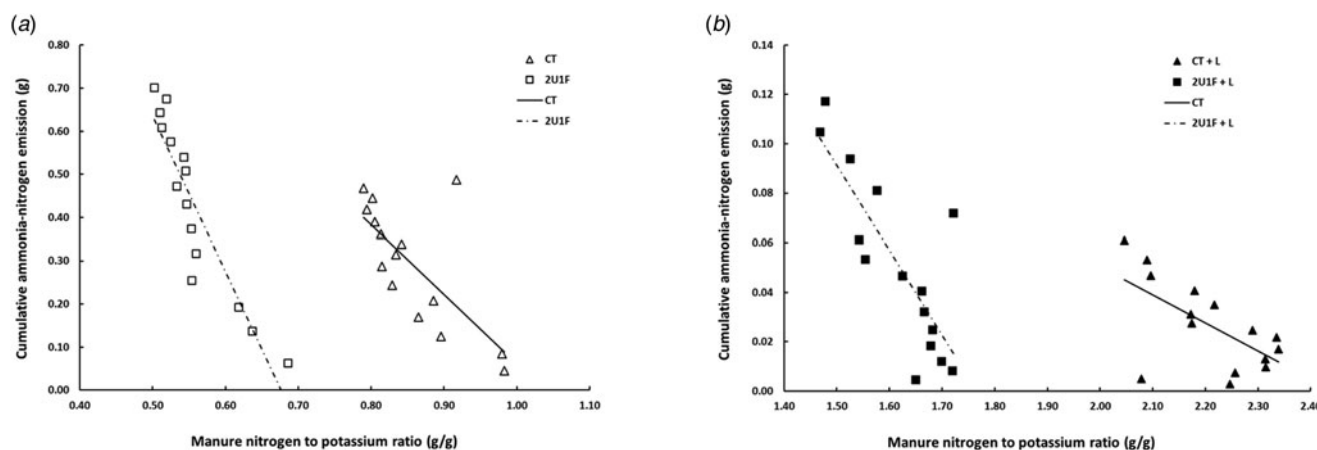
**Figure 3.** Cumulative ammonia-nitrogen emission (g) and the ratio of nitrogen and potassium (g/g) from 15 days dairy manure incubation. CT: UN:FN = 1:1; 2U1F: UN:FN = 2:1; CT + L: CT + Lignite; 2U1F + L: 2U1F + Lignite. CT used a manure with UN:FN = 1:1; 2U1F used a manure with UN:FN = 2:1; CT + L used a manure with UN: FN = 1:1 + 250 g lignite application; 2UIF + L used a manure with UN:FN = 2:1 + 250 g lignite application; UN:FN, urinary nitrogen: faecal nitrogen.

days may be caused by one outliner data on day 15; however, the relationships were strong in the other three treatments. Similar to the manure N: P ratio, the manure N: K ratio decreased with incubation days in four treatments, which were also approved by Moreira and Satter (2006) and Hristov et al. (2009).

The strong relationship between cumulative $NH_3$-N emission and manure $\delta^{15}N$ from the non-lignite group (combined CT and 2U1F treatments) were also reported by previous studies (Hristov et al., 2009; Lee et al., 2011; Cheng et al., 2013; Ayadi et al., 2015). Although Moreira and Satter (2006) used the manure N: P and N: K ratios to estimate $NH_3$-N emission, the moderate relationships were observed from 2U1F treatment in manure N: P ratio and from CT treatment in manure N: K ratio. The manure $\delta^{15}N$ could be the best biomarker used to estimate cumulative $NH_3$-N emission under normal $NH_3$-N concentrations.

Furthermore, the literature reviewed did not provide any references pertaining to the efficacy of N: P ratio, N: K ratio and $\delta^{15}N$ in mitigating $NH_3$ emissions at concentrations below 0.15 g per 24 h- the upper threshold applied in lignite application modalities within our experimental framework. No relationship was found between cumulative $NH_3$-N emission and manure $\delta^{15}N$ when combined CT + L and 2U1F + L treatments (Lignite group; $R^2$ = 0.02). This suggests that manure $\delta^{15}N$ may not serve as a preferable or accurate biomarker for estimating $NH_3$-N emissions in scenarios involving lignite. However, strong relationships were identified between cumulative $NH_3$-N emissions and manure N: P ratio under low $NH_3$-N concentrations. Conversely, the relationship between cumulative $NH_3$-N emissions and manure N: K ratio was weak in CT + L treatment but relatively strong in the 2U1F + L treatment. The manure N: P ratio emerged as a more accurate and effective biomarker for estimating $NH_3$-N emitted under conditions of low $NH_3$-N concentration. Therefore, further research is needed to validate the use of $\delta^{15}N$ to estimate $NH_3$ emissions under laboratory and field conditions, accounting for variations in cattle manure properties and climatic conditions.

The findings from this experiment have significant implications for understanding $NH_3$-N emissions from cattle manure, particularly in relation to lignite application. Understanding the underlying physical and biological processes leading to fractionation is crucial for accurate interpretation of isotopic data. Unravelling these mechanisms will enhance the precision of isotopic markers in assessing $NH_3$-N emissions. The experiment introduces the manure N: P ratio as a robust biomarker for $NH_3$-N emissions, outperforming other indicators under low $NH_3$-N concentrations. This has practical implications for developing more accurate estimation models, guiding manure management decisions, and potentially providing a more nuanced understanding of emission dynamics.

Future research should focus on the following objectives: (1) refining isotopic markers and improving their accuracy in estimating $NH_3$-N emissions requires a thorough understanding of the underlying physical and biological mechanisms of N fractionation; (2) it is necessary to validate the effectiveness of identified biomarkers (i.e., N: P ratio, N: K ratio, $\delta^{15}N$), under diverse climatic conditions, cattle manure properties, and management practices. These field studies will enhance the practical applicability of these biomarkers; (3) the utility and reliability of isotopic markers, particularly $\delta^{15}N$, should be examined under conditions of low $NH_3$-N emissions, as observed with lignite application treatments. Understanding the limitations of these markers will inform their appropriate use in various scenarios; and (4) creating integrated models that combine various biomarkers and environmental variables is crucial for accurately predicting $NH_3$-N emissions. This will contribute to the development of more comprehensive and precise strategies for managing and mitigating $NH_3$-N emissions from cattle manure.

## Conclusion

The experiment demonstrated a linear increase in manure $\delta^{15}N$ with rising $NH_3$ emissions and a decrease in the N: P ratio under specific treatments, indicating these markers complementary roles. The $\delta^{15}N$ was effective at high $NH_3$ concentrations, while the N: P ratio reliably estimated $NH_3$ emissions. These findings highlight the potential of lignite application as a mitigation strategy and suggest further research to optimize its use in livestock industries.

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